

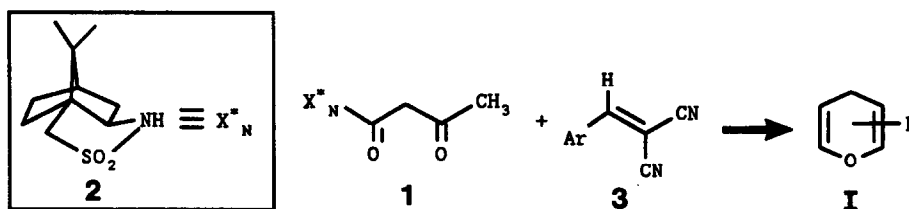
Asymmetric Alkylation of β -Ketoesters: Synthesis and Michael Additions of a Chiral Sultam-Derived Acetoacetyl Equivalent

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Abstract: The synthesis and Michael reaction of the chiral acetoacetyl equivalent **1** with arylidenemalononitriles **3** is reported.

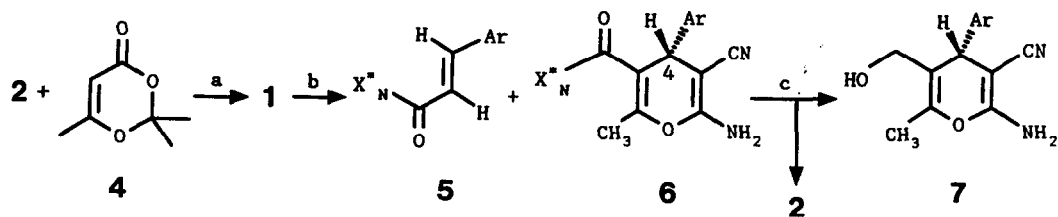
The asymmetric alkylation of β -ketoesters is a yet unsolved problem of great interest in organic synthesis; only recently several methodological developments have been reported.¹ In our current approach to this subject we have designed the new and chiral sultam-derived acetoacetyl equivalent **1** (Scheme 1), where chirality has been located at the carboxylic part of the β -ketoester moiety using the well known Oppolzer's sultam **2**.² In this letter we report the synthesis and asymmetric Michael reaction^{3,4} of compound **1** with arylidenemalononitriles **3**, eventually leading, after 1,4-addition and O-ring closure, to enantiomerically pure 4*H*-pyrans^{5,6} (**I**) (Scheme 1).



Scheme 1

The chiral reagent **1**⁷ has been synthesized from 2,2,6-trimethyl-4*H*-1,3-dioxin-4-one **4** and 10,2-bornanesultam **2**² in 86% yield following the standard protocol⁸ (Scheme 2). Initial exploratory experiments directed to find the best conditions for the reaction of **1** and **3** (Ar = C₆H₅)⁹ gave unsuccessful results using titanium tetrachloride (CH₂Cl₂, -78°C → r.t.)¹⁰, boron trifluoride etherate (CH₂Cl₂, -78°C → r.t.), dimethyl-*t*-butylsilyl triflate (CH₂Cl₂, NEt₃; TiCl₄)¹¹ or LDA (-78°C → r.t.) as promoters or catalysts; using sodium hydroxide under PTC conditions¹², in addition to recovered starting materials, sultam **2**, benzaldehyde and *N*-(*E*-cinnamoyl)-10,2-bornanesultam **5**¹³ (Ar = C₆H₅) were detected. This compound, as the *E* isomer only, was obtained in 74% yield when sodium hydride (THF; 0°C → r.t.) was used as base. No 4*H*-pyran was observed. Finally, using piperidine as catalyst in toluene,^{5,14} compound **1** reacted with **3** giving, after

Michael addition and *O*-ring closure, the expected 4*H*-pyrans **6**⁷, in yields and diastereomeric excesses as shown in Table, and traces of the corresponding *N*-(*E*-3-arylpropenoyl)-10,2-bornanesultams (**5**). From the results in Table we conclude that we have obtained consistent and good values for the d.e.,s ($\approx 60\%$) in the asymmetric addition of our *N*-acetoacetyl sultam **1** to the Michael acceptors **3**. These data are also independent of the nature and position of the substituent attached to the phenyl ring in **3**. In the ¹H NMR spectra of these diastereomeric mixtures, major and minor isomers **6** showed clearly resolved signals for H-4 at 4.7-4.6 ppm and 4.3-4.2 ppm, respectively. Unfortunately, we were unable to separate these isomers by flash chromatography;¹⁵ only in cases **6a** and **6f** (see Table) we could obtain pure major and minor isomers by recrystallization. After careful analysis of the ¹H NMR spectra (see above) of diastereomers **6**, it is clear that in all these cases the major isomer has the same absolute configuration at the new stereocenter (C-4). This stereochemistry has been established as *R* by X-ray diffraction analysis of minor **6a**-(*S*).¹⁶



a: toluene, 150°C, 82%; b: **3**, piperidine, toluene, r.t.; c: LiAlH₄, ether, reflux.

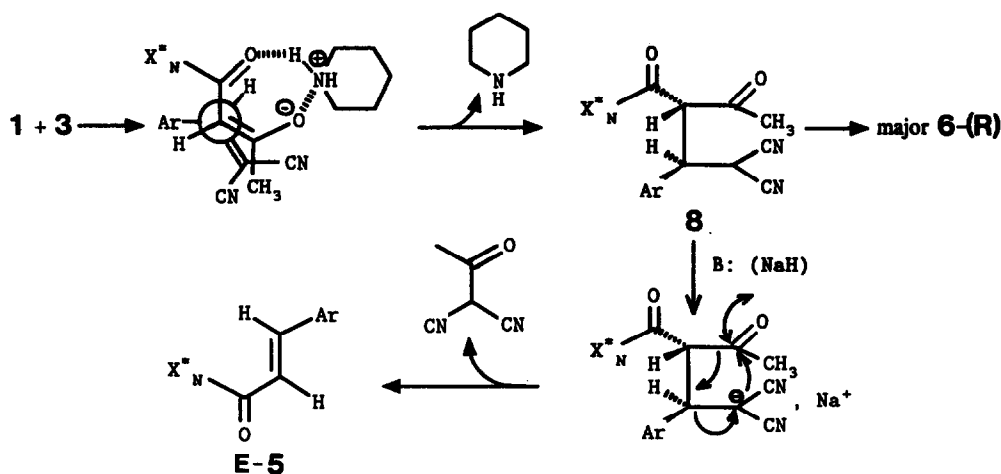
Scheme 2

Table. New 4*H*-pyrans **6** and **7** prepared.

	Ar	6		7
		Yield ¹ (%)	d.e. ² (d.e.) ³	Yield ¹ (%)
a	C ₆ H ₅	65	60 (>99)	73
b	<i>o</i> -CH ₃ -C ₆ H ₄	62	60	72
c	<i>p</i> -CH ₃ -C ₆ H ₄	42	70	59
d	<i>m</i> -NO ₂ -C ₆ H ₄	85	70	40
e	<i>p</i> -Br-C ₆ H ₄	91	60	41
f	<i>p</i> -Cl-C ₆ H ₄	64	70 (>99)	42

¹ After chromatography. ² Determined by ¹H NMR (300 MHz) on the crude reaction mixtures. ³ After recrystallization.

Finally, after extensive search for optimal conditions¹⁷, pure major isomers **6a**, **6f** and the diastereomeric mixtures **6b-e** were treated with lithium aluminium hydride¹⁴, providing sultam **2** (75-95% yield) and the enantiomerically pure alcohols **7a**⁷, **7f**⁷ or the enantiomerically enriched alcohols **7b-e** in reasonable yield (see Table).



Scheme 3

The formation of major **6(R)** isomers coupled to the exclusive formation of *E-5* isomers (traces) during the Michael addition can be explained as shown in Scheme 3. We can assume that the α -unsubstituted acylsultam reacts in a conformation where the carbonyl is *anti* to the SO_2 group^{2,18} (also confirmed by X-Ray analysis of compound **1**¹⁶) and *s-cis* to the $\text{C}_\alpha\text{C}_\beta$ bond due to a presumed internal chelate between the piperidinium cation and the 1,3-dicarbonyl group. The face differentiation is then dictated by the Michael addition to the arylidenemalononitrile from the less hindered bottom face to generate intermediate **8**, that smoothly gives **5** (traces) and **6**. This particular behaviour has also been observed during the 1,4-hydride addition to enoylsultams.¹⁹ As shown, the exclusive formation of *E-5* is based on an irreversible intramolecular addition-elimination process. Note that, not surprisingly, in the presence of a stronger base, as sodium hydride (see above), formation of *E-5* (Ar = C_6H_5) is preferred to the *O*-ring closure.

The results reported here add to the large panoply of asymmetric syntheses with camphor sultam derivatives and give a clear picture of the power of this chiral auxiliary in order to induce good selectivity in Michael additions. Compound **1** is thus an efficient acetoacetyl chiral equivalent of potential interest in asymmetric synthesis. In addition, with these results, we have improved, in yields and diastereoselectivity, our former experiments⁶ in the asymmetric synthesis of multiply functionalized *4H*-pyrans. Work is now in progress in order to exploit and expand the synthetic usefulness of intermediate **1** and will be reported in due course.

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REFERENCES AND NOTES

- Ando, K.; Takemasa, Y.; Tomioka, K.; Koga, K. *Tetrahedron* **1993**, *49*, 1579, and reference 4 cited therein.
- Reviews: (a) Oppolzer, W. *Pure Appl. Chem.* **1990**, *62*, 1241. (b) Oppolzer, W. *Tetrahedron* **1987**, *43*, 1969.
- Perlmutter, P. *Conjugate Addition Reactions in Organic Synthesis*; Tetrahedron Organic Chemistry Series No 9, Pergamon Press: Oxford, 1992. For review see: Lee, V.J. In *Comprehensive Organic Synthesis*; Trost, B.M.; Fleming, I., Pergamon Press: Oxford, 1991; Vol 4, Chapter 1.2.
- For previous reviews on stereoselective conjugate additions see: (a) Rossiter, B.E.; Swingle, N.M. *Chem. Rev.* **1992**, *92*, 771. (b) Schmalz, H.G. In *Comprehensive Organic Synthesis*; Trost, B.M.; Fleming, I., Pergamon Press: Oxford, 1991; Vol 4, Chapter 1.5. (c) Tomioka, K.; Koga, K. In *Asymmetric Synthesis*; Morrison J.D., Academic Press, New York, 1983; Vol 2, p 201.
- Soto, J.L.; Seoane, C.; Martín, N.; Quinteiro, M. *Heterocycles* **1984**, *22*, 1.
- This is also a complementary approach to our recently described asymmetric Michael addition of malononitrile to chiral α -acylacrylates: González, R.; Martín, N.; Seoane, C.; Marco, J.L.; Albert, A.; Cano, F.H. *Tetrahedron Lett.* **1992**, *33*, 3809.
- All new compounds have shown good analytical and spectroscopic data: **1**: m.p. 78-81°C; $[\alpha]_D^{25}$ -77.5° (c 1.06, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ : 4.00 (d, 1H, *J*=16.8 Hz, CH₂CO), 3.87 (dd, 1H, *J*=7.8 and 4.8 Hz, CHN), 3.66 (d, 1H, *J*=16.8 Hz, CH₂CO), 3.49 (d, 1H, *J*=13.8 Hz, CH₂SO₂), 3.42 (d, 1H, *J*=13.8 Hz, CH₂SO₂), 2.23 (s, 3H, CH₃CO), 2.30-1.85 (5H), 1.45-1.32 (2H), 1.13 (s, 3H, CH₃), 0.95 (s, 3H, CH₃); **6a** (major isomer): m.p. 185-187°C; $[\alpha]_D^{25}$ -165.6° (c 0.9, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ : 7.33-7.18 (5H, aromatic), 4.64 (d, 1H, *J*=1.2 Hz, H-4), 4.50 (br s, 2H, NH₂), 3.82 (dd, 1H, *J*=7.8 and 4.8 Hz, CHN), 3.48 (d, 1H, *J*=13.8 Hz, CH₂SO₂), 3.38 (d, 1H, *J*=13.8 Hz, CH₂SO₂), 1.99 (d, 3H, *J*=1.2 Hz, CH₃), 1.93-1.72 (4H), 1.42-1.22 (3H), 1.09 (s, 3H, CH₃), 0.94 (s, 3H, CH₃); ¹³C NMR (75 MHz, CDCl₃) δ : 166.45 (CO), 157.45 (C2), 149.75 (C6), 141.13, 128.49, 127.95, 127.54 (aromatic), 118.91 (CN), 110.35 (C5), 64.29 (CHN), 61.31 (C3), 52.87 (CH₂SO₂), 48.14, 47.60 (2 C), 44.11 (CH), 40.34 (C4), 37.35, 32.45, 26.39 (3 CH₂), 20.44, 19.68, 17.25 (3 CH₃); **7a**: m.p. 95-98°C; $[\alpha]_D^{25}$ +25° (c 1.3, CH₂COCH₃); ¹H NMR (300 MHz, CD₂COCD₂) δ : 7.35-7.20 (5H, aromatic), 5.86 (br s, 2H, NH₂), 4.21 (d, 1H, *J*=1.2 Hz, H-4), 4.11 (d, 1H, *J*=12.3 Hz, CH₂OH), 3.60 (d, 1H, *J*=12.3 Hz, CH₂OH), 1.97 (d, 3H, *J*=1.2 Hz, CH₃); ¹³C NMR (75 MHz, CDCl₃) δ : 160.51 (C2), 144.23 (C6), 145.30, 129.26, 128.77, 127.67 (aromatic), 120.46 (CN), 113.36 (C5), 59.63 (C3), 59.03 (CH₂OH), 40.53 (C4), 15.44 (CH₃); **7f**: m.p. 111-113°C; $[\alpha]_D^{25}$ +37.6° (c 1.23, CH₂OH); ¹H NMR (300 MHz, CDCl₃) δ : 7.30 (td, 2H, aromatic), 7.18 (td, 2H, aromatic), 4.56 (br s, 2H, NH₂), 4.18 (d, 1H, *J*=0.9 Hz, H-4), 4.07 (d, 1H, *J*=12.3 Hz, CH₂OH), 3.74 (d, 1H, *J*=12.3 Hz, CH₂OH), 1.98 (d, 3H, *J*=0.9 Hz, CH₃); ¹³C NMR (75 MHz, CDCl₃) δ : 158.71 (C2), 144.61 (C6), 141.58, 133.14, 129.22, 128.90 (aromatic), 119.44 (CN), 111.28 (C5), 60.46 (C3), 59.71 (CH₂OH), 39.34 (C4), 15.34 (CH₃).
- Clemens, R.J.; Hyatt, J.A. *J. Org. Chem.* **1985**, *50*, 2431.
- (a) For the Michael addition of chiral malonate derivatives to α,β -unsaturated carbonyl compounds see: Mukaiyama, T.; Hirako, Y.; Takeda, T. *Chem. Lett.*, **1978**, 461. (b) For the α -alkylation of a chiral cyanoacetate ester see also: Cativiela, C.; Díaz de Villegas, M.D.; Gálvez, J.A. *Tetrahedron: Asymmetry* **1992**, *3*, 1141. (c) For the intramolecular Michael reaction of a chiral β -ketoester, see: Stork, G.; Saccomano, N.A. *Nouv. J. Chim.* **1986**, *10*, 677.
- Oppolzer, W.; Rodriguez, I.; Blagg, J.; Bernardinelli, G. *Helv. Chim. Acta* **1989**, *72*, 123.
- Oppolzer, W.; Starkeman, C.; Rodriguez, I.; Bernardinelli, G. *Tetrahedron Lett.* **1991**, *32*, 61.
- Oppolzer, W.; Moretti, R.; Thomi, S. *Tetrahedron Lett.* **1989**, *30*, 1009.
- Kocienski, P.; Thomi, C. *Synthesis* **1992**, 582.
- General procedure for the synthesis of **6** and **7**: Compound **1** (1.0 equiv) was dissolved in dry toluene, and piperidine (four drops) plus the appropriate arylidenemalononitrile **3** (1.5 equiv) were added. The solution was stirred at r.t. for 24-48 h. The solvent was removed and the residue submitted to FC giving compound **6**. A solution of **6** (1.0 equiv) in THF/ether (1:5, 18 mL/0.5 mmol) was added to a stirred suspension of LiAlH₄ (2.5 equiv) in dry ether (4 mL/mmol) at 0°C. Stirring (0°C, 3h), addition of saturated aqueous NH₄Cl, extraction with ether, drying, concentration and FC gave the recovered sultam **2** (75-95% yield) and the alcohol **7**.
- Still, W.C.; Kahn, M.; Mitra, A. *J. Org. Chem.* **1978**, *43*, 2923.
- Personal communication (F.H. Cano, U.E.I. Cristalografía, Instituto de Química Física "Rocasolano" (CSIC), Serrano 116, 28006-Madrid, Spain).
- Oppolzer, W.; Lienard, P. *Helv. Chim. Acta* **1992**, *75*, 2572.
- Curran, D.P.; Kim, B.H. *Tetrahedron* **1993**, *49*, 293.
- Oppolzer, W.; Poli, G.; Starkemann, C.; Bernardinelli, G. *Tetrahedron Lett.* **1988**, *29*, 3559.

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